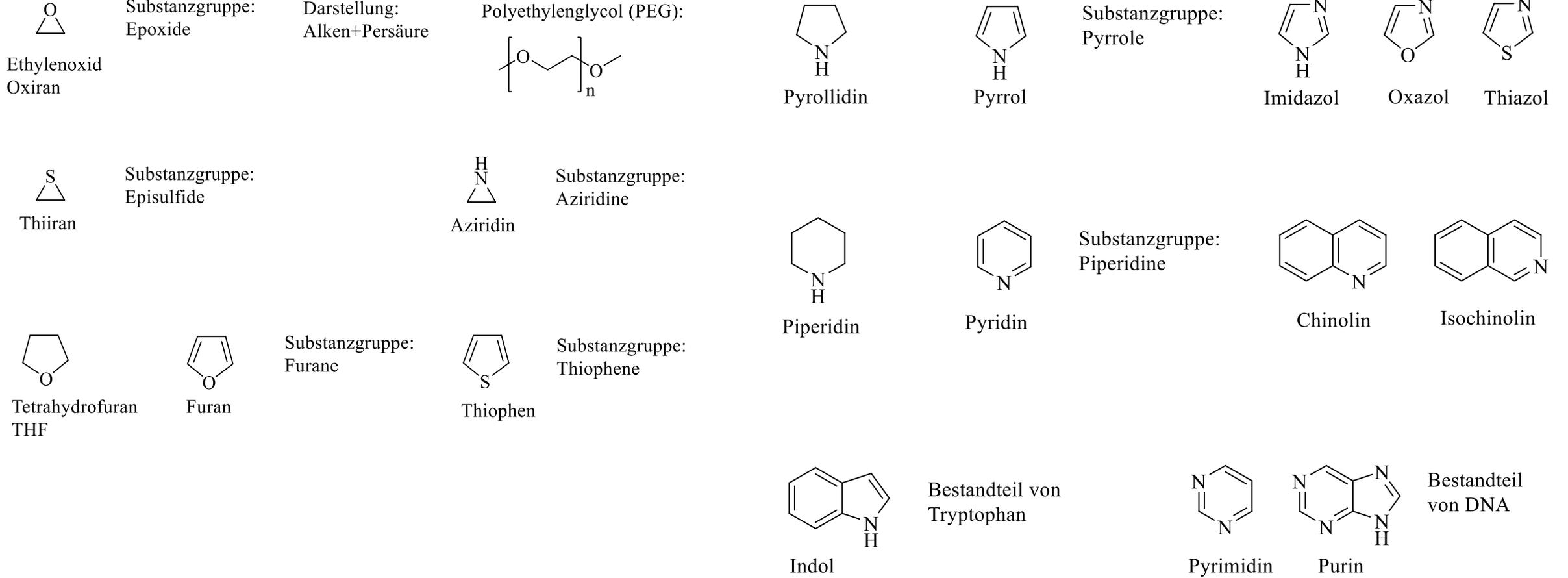


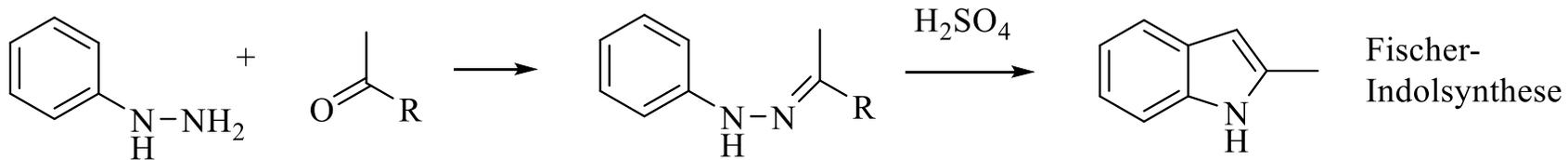
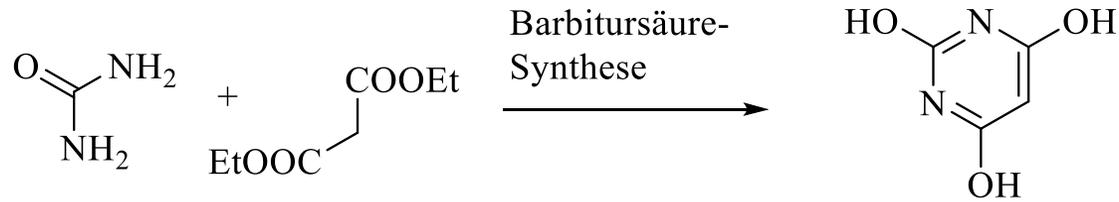
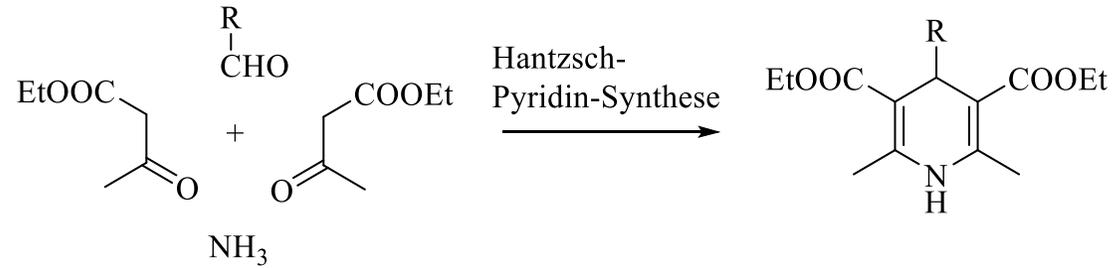
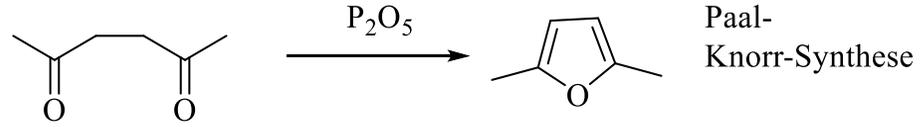
Heterocyclen

Wichtige Heterocyclen



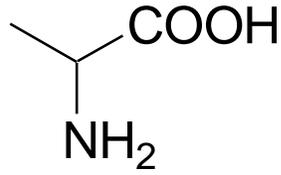
Heterocyclen

Einige wichtige Synthesen

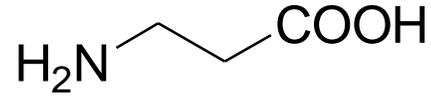


Naturstoffe

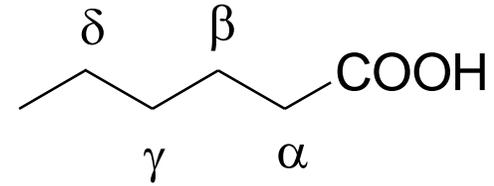
Aminosäuren



α -Alanin (chiral)

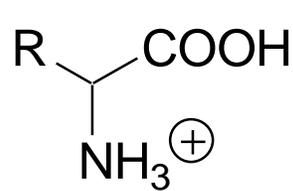


β -Alanin (achiral)

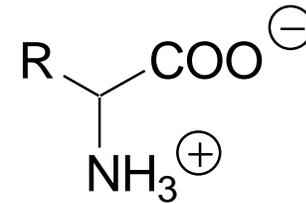
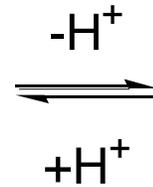


Chemische Eigenschaften

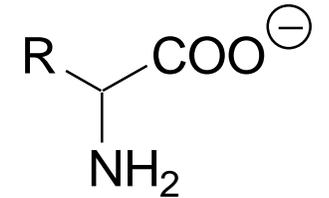
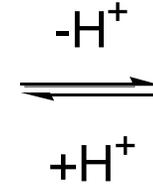
Aminosäuren sind meist kristalline Feststoffe und liegen als Zwitterionen vor. Den pH-Wert, bei dem eine Aminosäure in der (nach außen ungeladenen) zwitterionischen Form vorliegt wird Isoelektrischer Punkt (IP) genannt. Der IP ist für jede Aminosäure charakteristisch.



im Sauren



"neutral"
pH = IP



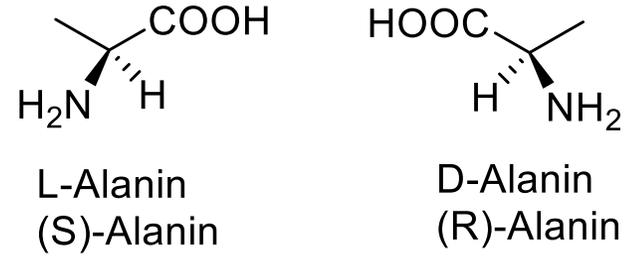
im Basischen

Aminosäuren

Einteilung der Aminosäuren:

Chiralität:

In der Natur fast ausschließlich L-AS

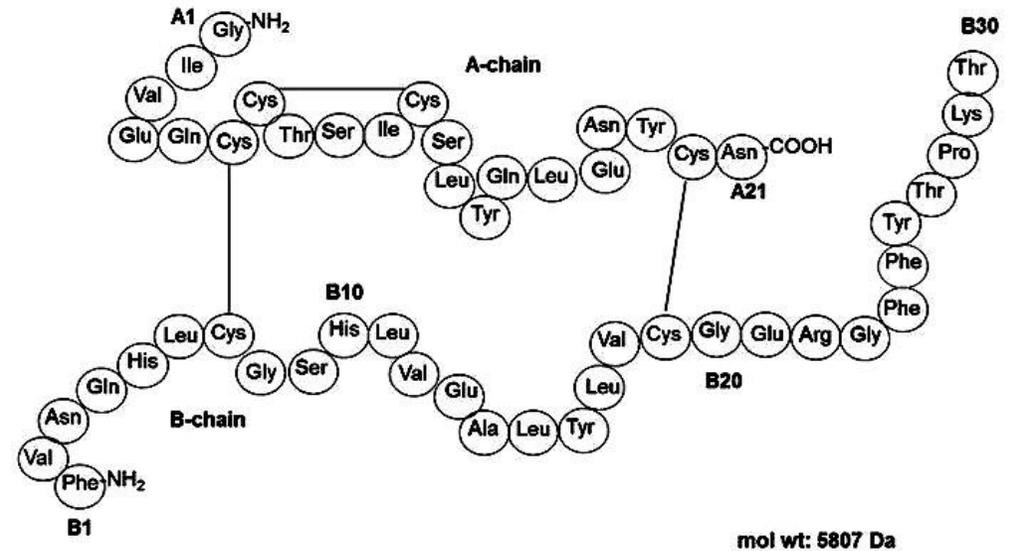


Proteinogene Aminosäuren:

Aminosäuren, die in Proteine eingebaut werden.

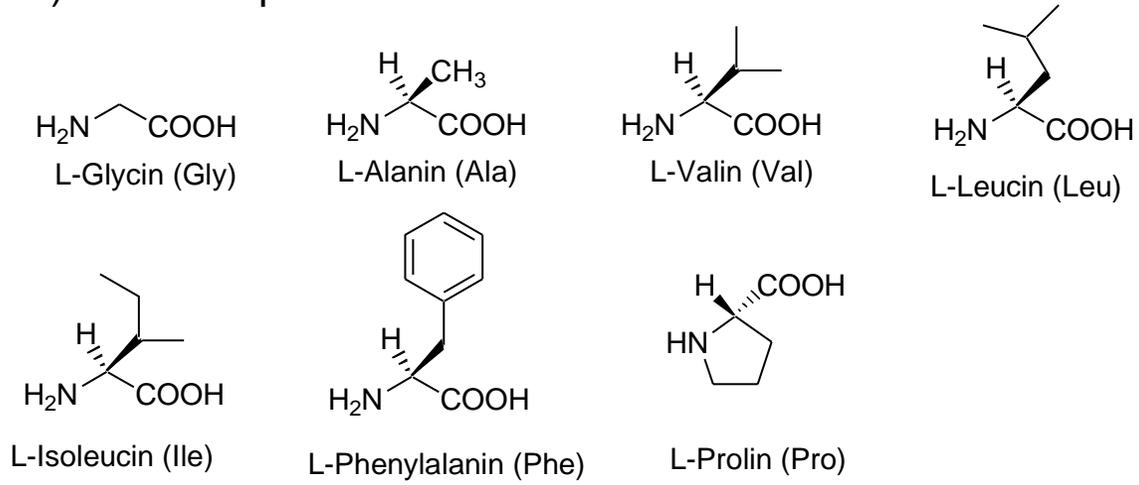
Essentielle Aminosäuren: Aminosäuren, die mit der Nahrung aufgenommen werden müssen (Mensch).

L-Valin, L-Leucin, L-Isoleucin, L-Phenylalanin,
L-Threonin, L-Methionin, L-Tryptophan, L-Lysin

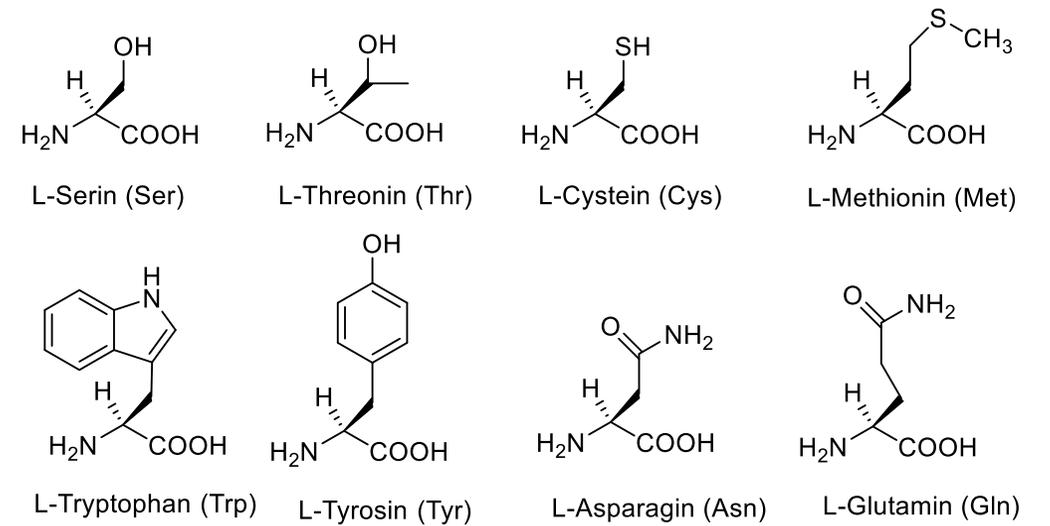


Proteinogene Aminosäuren

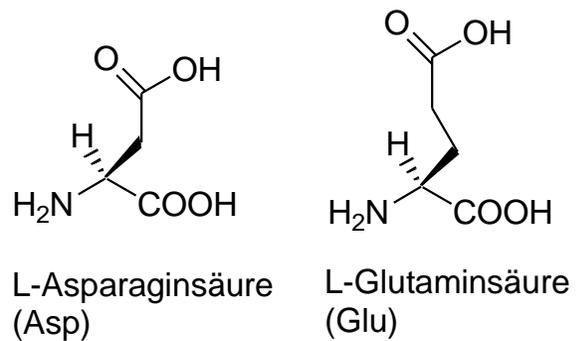
a) AS mit unpolarem Rest:



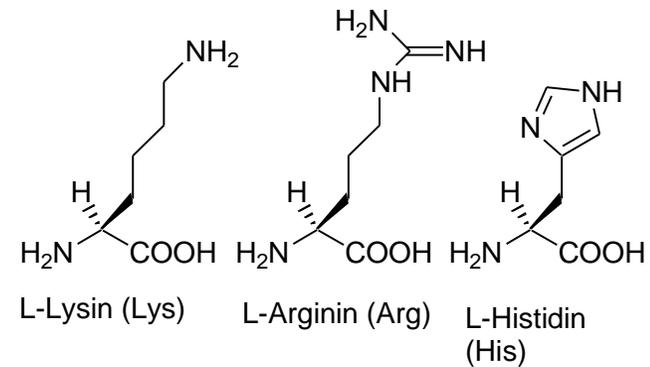
b) AS mit polarem, nicht ionisierbarem Rest:



c) polare saure AS mit ionisierbarem Rest:



d) polare basische AS mit ionisierbarem Rest:



Aminosäuresynthesen

Bedarf an Aminosäuren

- Jahresproduktion ca. 1,6 Millionen Tonnen
- Glutaminsäure: 650 000 Tonnen
- D,L-Methionin: 450 000 Tonnen
- L-Lysin: 450 000 Tonnen
- L-Threonin: 30 000 Tonnen

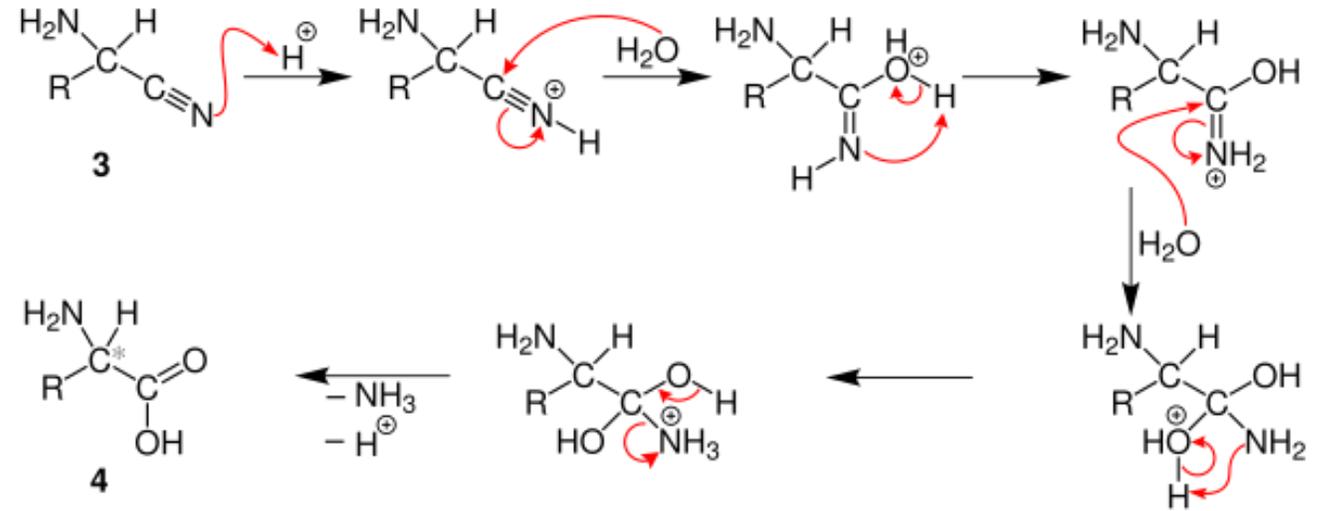
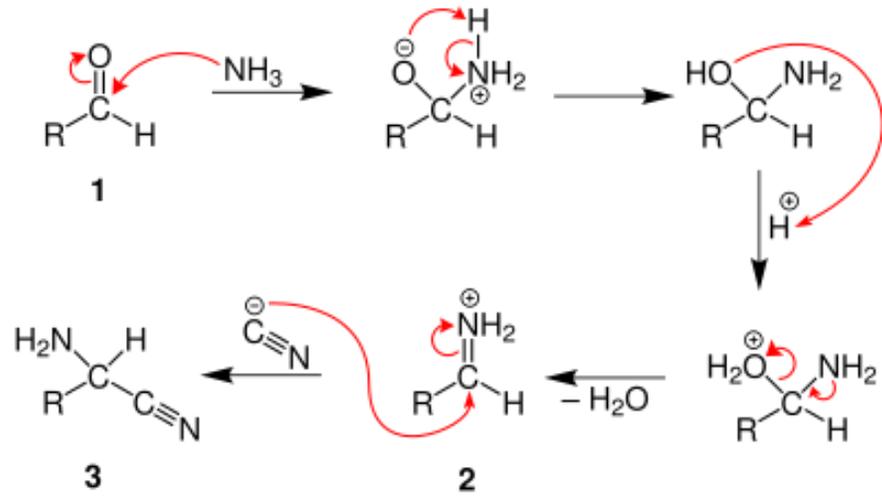
Verwendung:

- Lebensmittelindustrie: Geschmacksverstärker, Süßstoffe (Glu, Gly, Cys)
- Futtermittelindustrie: Tierernährung (Met, Lys, Trp, Thr)
- Pharmaindustrie: Medikamente, Infusionslösungen (Acetyl-Cys, L-DOPA)
- Kosmetika, Pflanzenschutzmittel, Stabilisatoren für PVC, Dispersionshilfen, Hilfsmittel in der Galvanotechnik

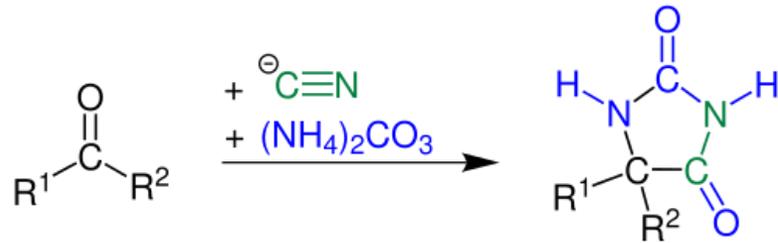
Herstellungsmethoden

- Extraktionsmethode (Cystein Tyrosin Prolin aus Kollagen)
- Enzymatische Methoden (L-Leucin, L-Methionin)
- Fermentationsmethode (L-Glutaminsäure aus D-Glucose)
- Chemische Synthesen

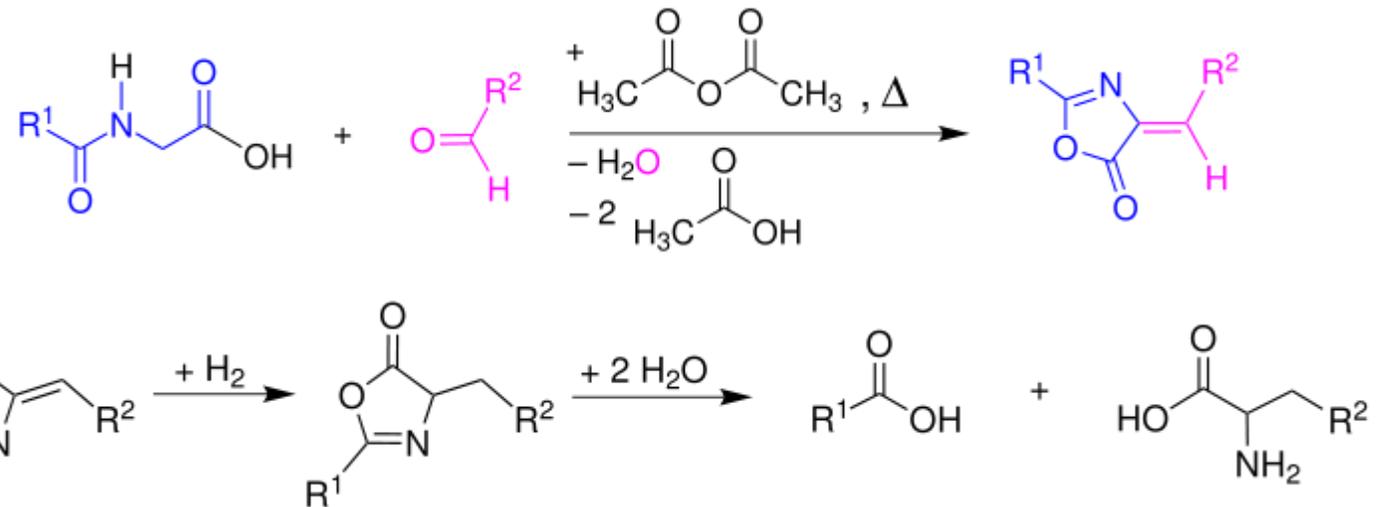
Strecker-Synthese



Bucherer-Bergs-Synthese

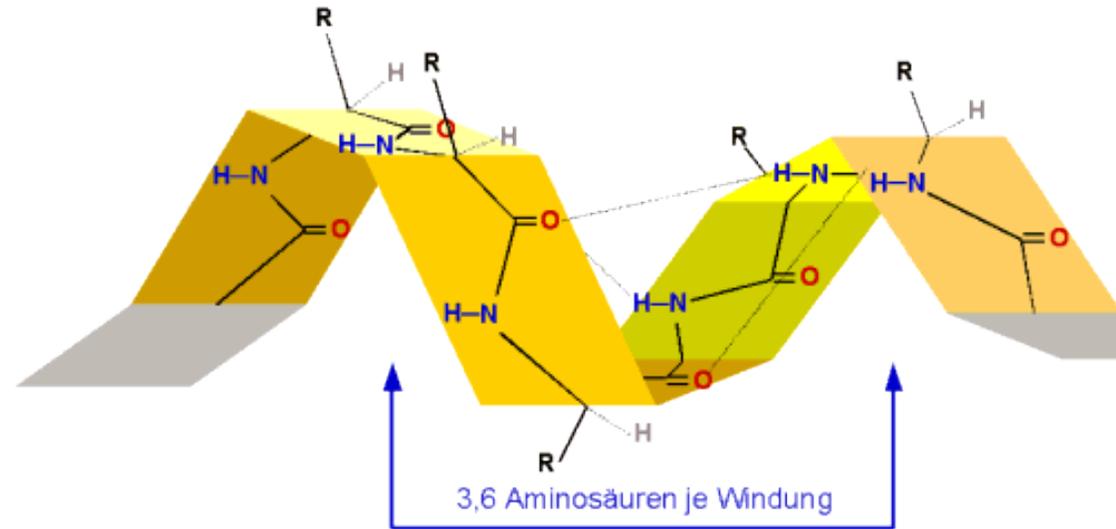
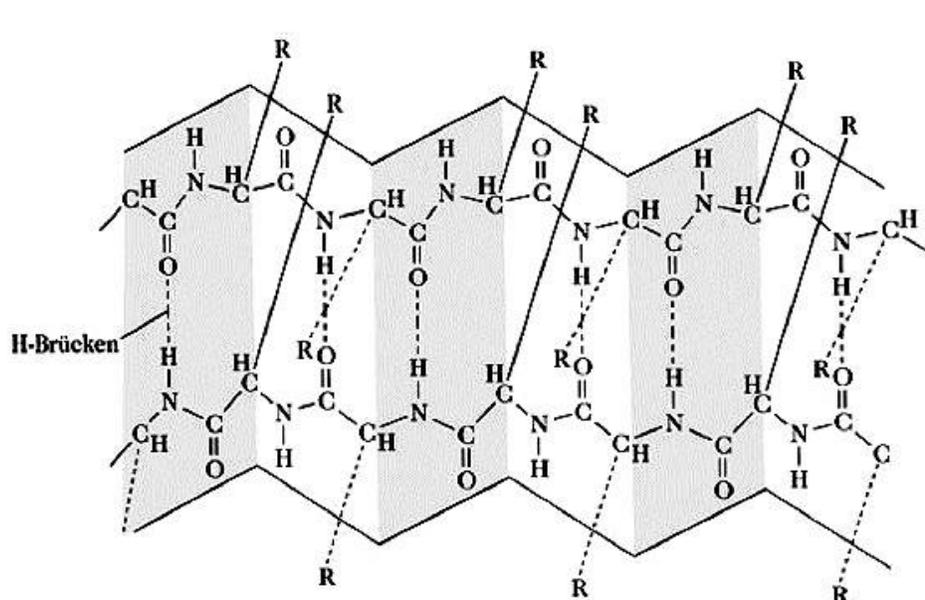


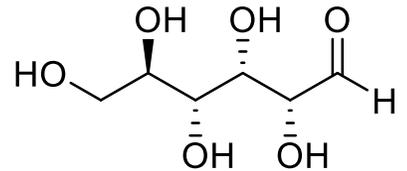
Erlenmeyer-Synthese (Azlacton-Synthese)



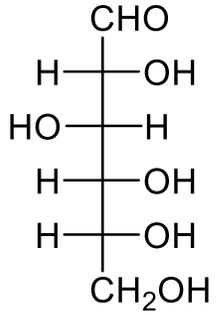
Strukturen von Proteinen:

1. Primärstruktur: Aminosäuresequenz
2. Sekundärstruktur: räumliche Anordnung der Kette (Flatblatt, Helix)
nur peptidisches Rückgrat, nicht räuml.
Anordnung der AS Seitenketten
3. Tertiärstruktur: Gestalt des Proteins (räuml. Anordnung aller Atome)
4. Quartärstruktur: Aggregate aus mehreren Proteinen (Dimere, Trimere, etc.)

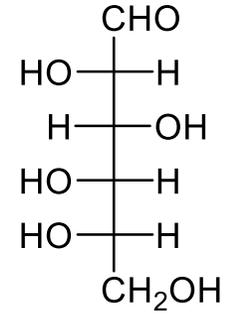




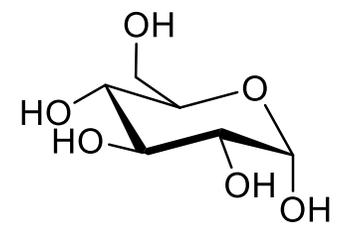
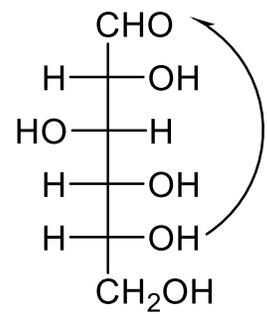
(2*R*,3*S*,4*R*,5*R*)-2,3,4,5,6-Pentahydroxyhexanal



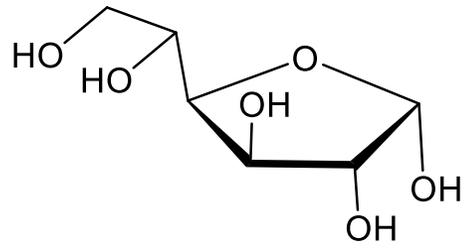
D-Glucose



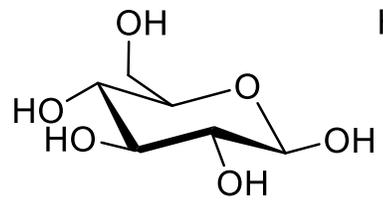
L-Glucose



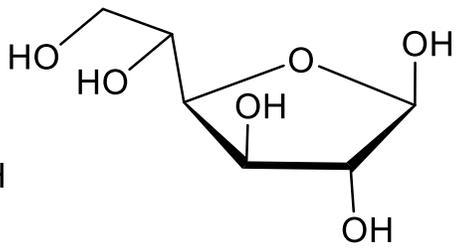
α -D-Glucopyranose



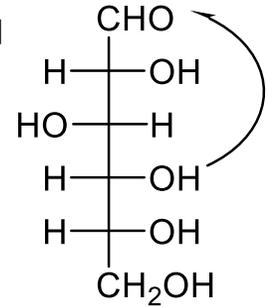
α -D-Glucofuranose



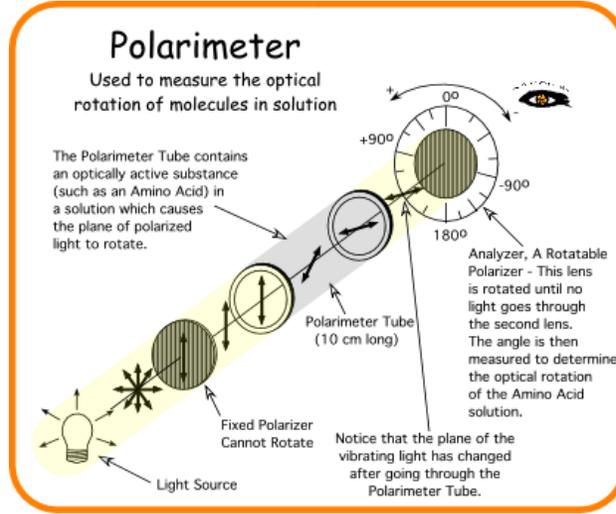
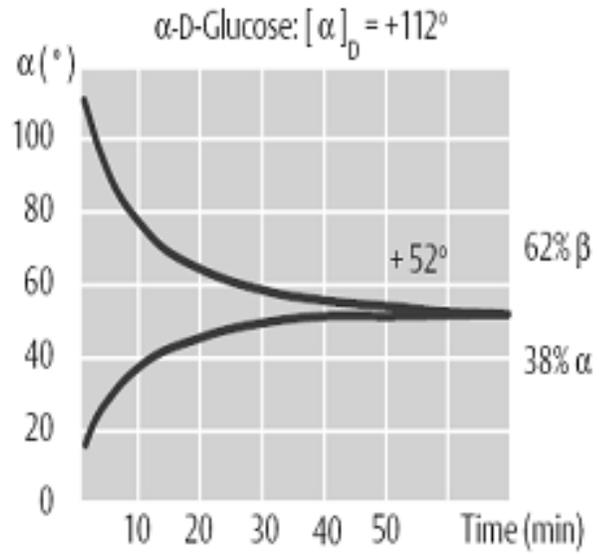
β -D-Glucopyranose



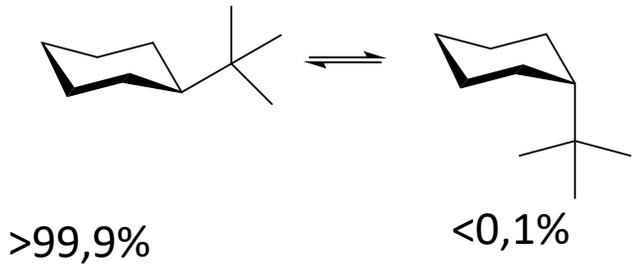
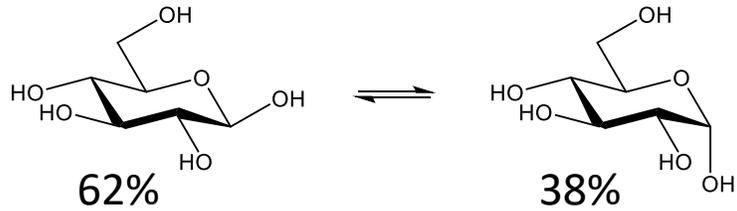
β -D-Glucofuranose



Mutarotation

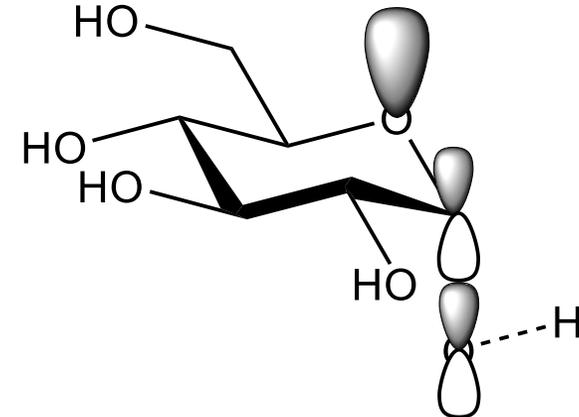


β -D-Glucose: $[\alpha]_D = +19^\circ$



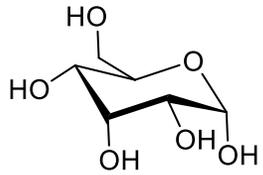
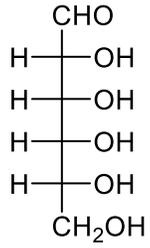
Anomerer Effekt

Elektronegative Substituenten an C1 Stabilisieren das α -Anomer

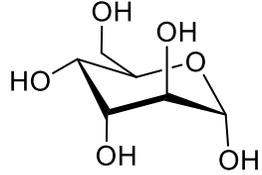
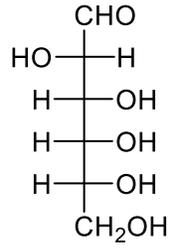


Hexosen / Pyranosen

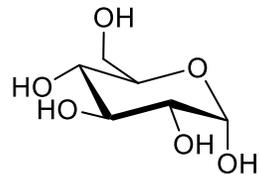
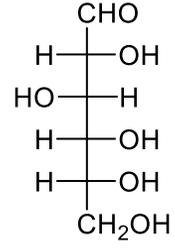
D-Allose



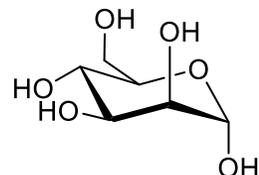
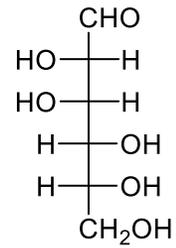
D-Altrose



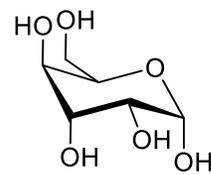
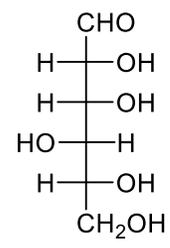
D-Glucose



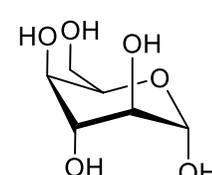
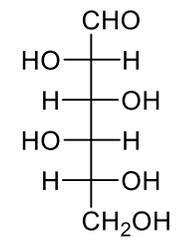
D-Mannose



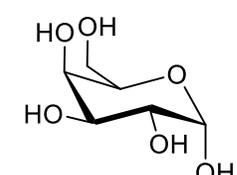
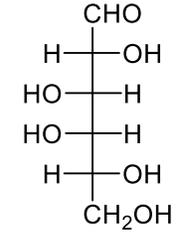
D-Gulose



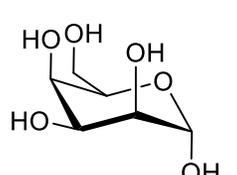
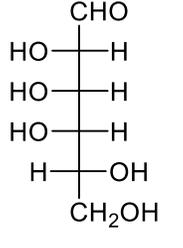
D-Idose



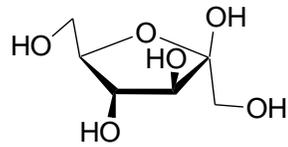
D-Galactose



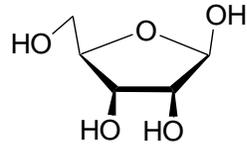
D-Talose



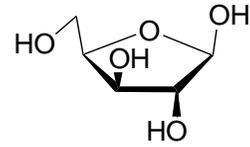
Furanosen



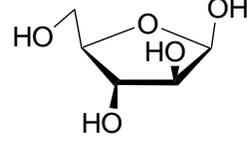
Fructose



Ribose



Xylose

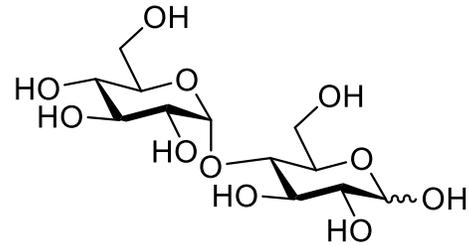


Arabinose

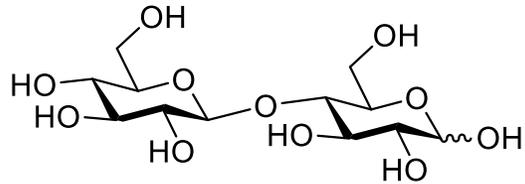
Saccharide, Glycoside

1,4-, 1,6-Verknüpfung, Saccharose, Lactose, Maltose, Isomaltose, Cyclodextrine, Bildung und Hydrolyse von Sacchariden

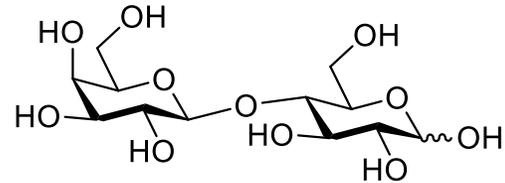
Maltose
4-O-(α -D-Glucopyranosyl)-D-glucopyranose



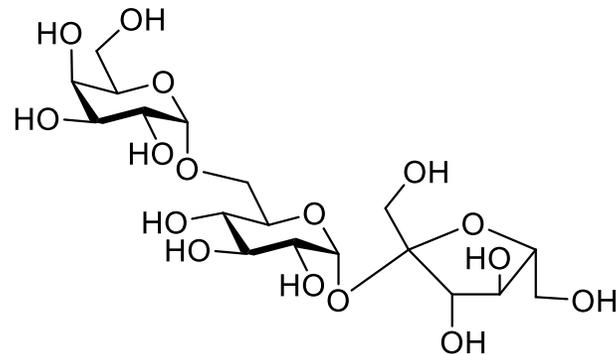
Cellobiose
4-O-(β -D-Glucopyranosyl)-D-glucopyranose



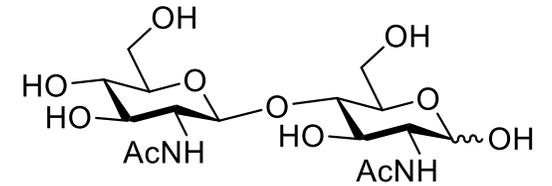
Lactose (Milchzucker)
4-O-(β -D-Galactopyranosyl)-D-glucopyranose



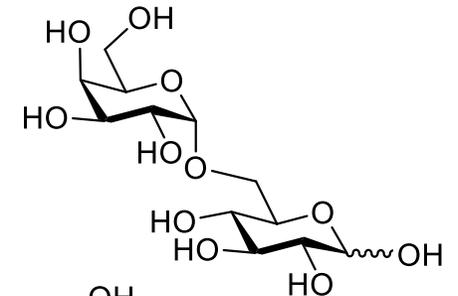
Raffinose (Rübenzucker)



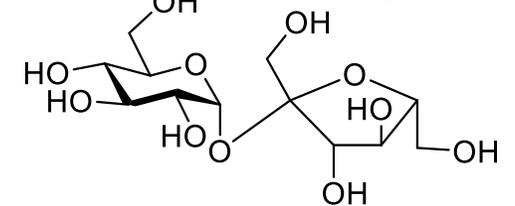
Chitobiose (Chitin)



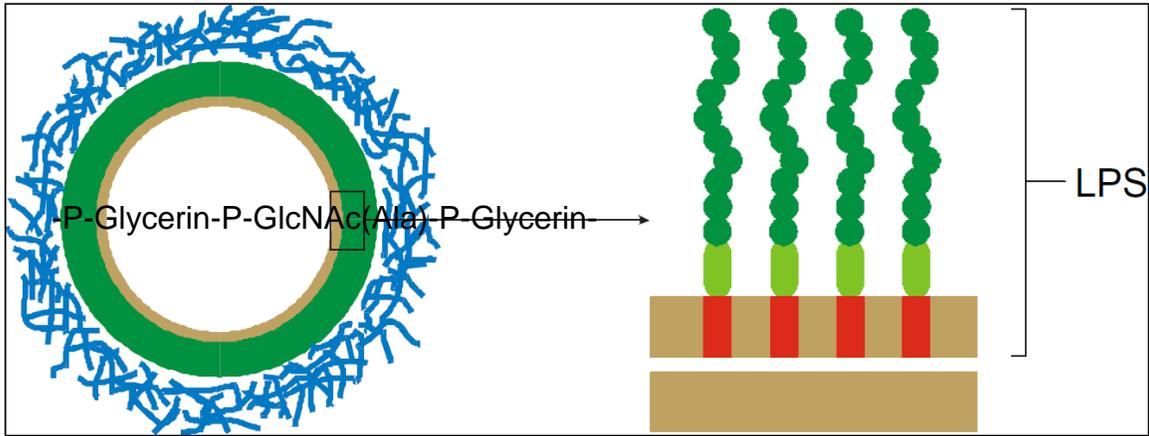
Melibiose



Saccharose (Rohrzucker)



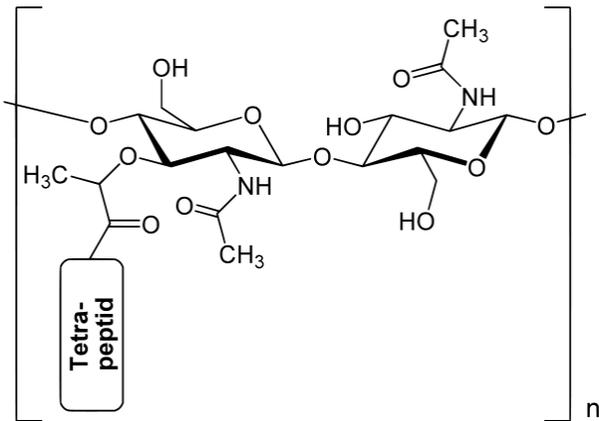
Bakterienoberfläche



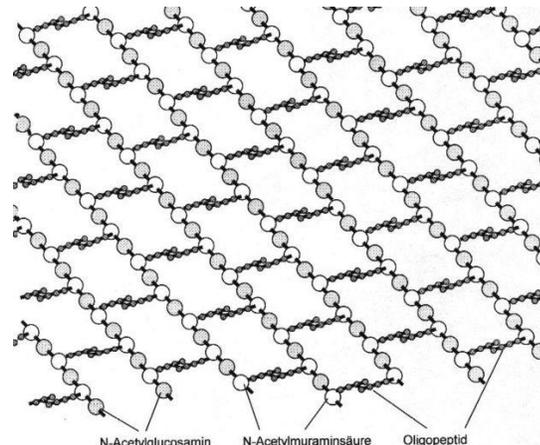
Schematische Darstellung der bakteriellen Polysaccharide

äußere Membran (braun), Lipopolysaccharid LPS (grün), Exopolysacchararid EPS (blau) Lipid A (rot), core Polysaccharid (hellgrün) und O-Antigen (dunkelgrün). EPS ist sowohl in Gram-negativen wie auch Gram-positiven Bakterien anzutreffen.

Murein

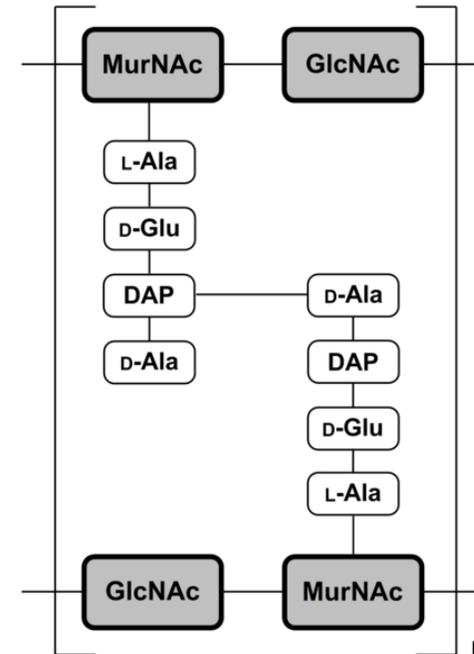
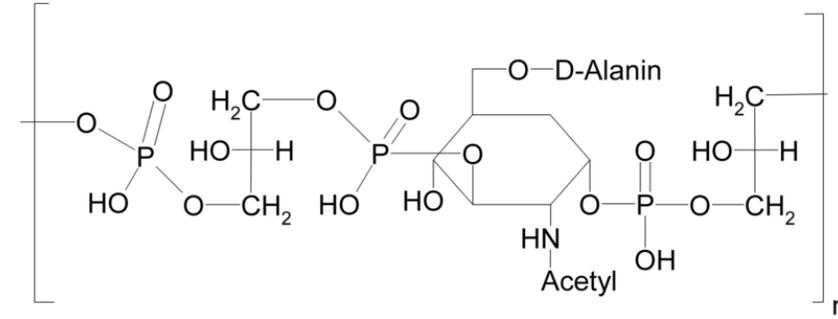


N-Ac-Muraminsäure GlcNAc

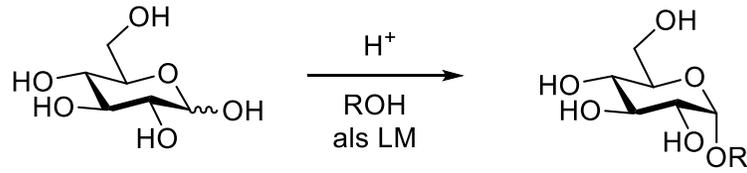


Peptidoglykan der Bakterien-Zellwand

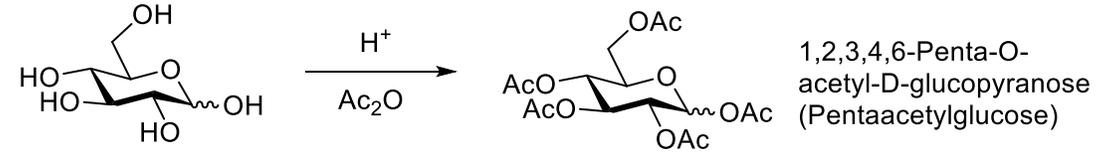
Teichonsäure



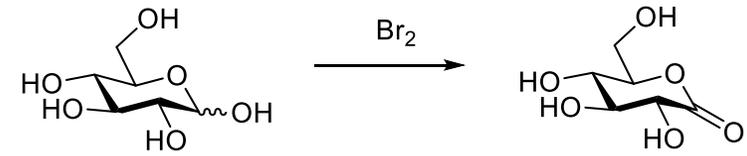
Reaktionen der Zucker



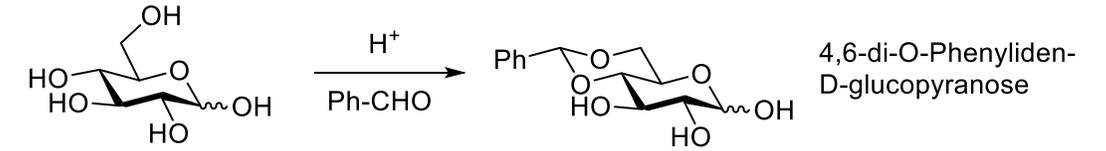
Methyl- α -D-glucopyranosid



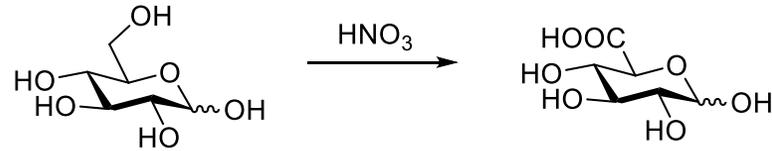
1,2,3,4,6-Penta-O-acetyl-D-glucopyranose (Pentaacetylglucose)



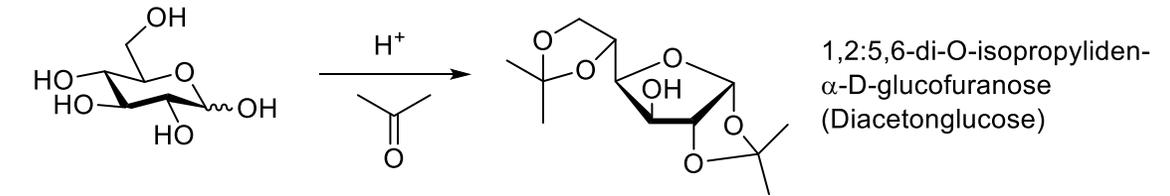
D-Glucono- δ -lacton



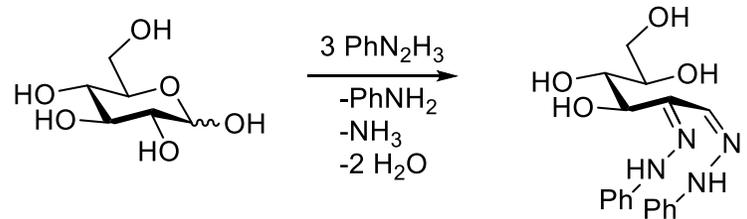
4,6-di-O-Phenyliden-D-glucopyranose



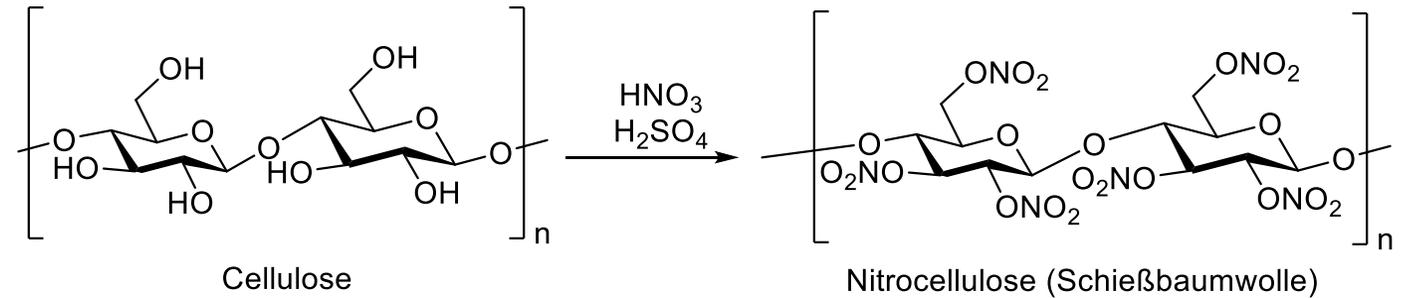
D-Glucuronsäure



1,2:5,6-di-O-isopropyliden- α -D-glucofuranose (Diacetonglucose)



Osazon



Cellulose

Nitrocellulose (Schießbaumwolle)